名　　稱	分　子　式	備注	存放地點
A:小於 1/2　B:快完　C:潮解、結塊　甲類先驅物			
A			
Acetic acid (99-100%)	CH_3COOH		9
Acetic acid glacial		2012/6/26	9
Acetic acid, sodium salt, 99%			1
Acetic acid, glacial p.a. 99.8%	$C_2H_4O_2$	B	9,P,P
Activated charcoal	C		P
Acetone p.a.	C_3H_6O	C	1
Acetonitrile (99+%)	C_2H_3N	A	1
Agar		2012.11 A	2
Agefloc A-50			
Agefloc WT-40			
Albumin, bovine serum			P
Alcohol 95%			
AlOOH			1
Alumina polishing powder	200 mesh		P,P
Aluminum chloride, anhydrous powder 98.5%	$AlCl_3$		1
Aluminum chloride hexahydrate 99% pure			P
	$AlCl_3 \bullet 6H_2O$	2012/2/3	
Aluminium nitrate nonahydrate GR	$Al(NO_3)_3 \bullet 9H_2O$	C	2
Aluminium nitrate nonahydrate, 99%, for analysis	$AlN_3O_9.9H_2O$	C	2
Aluminium hydroxide	$Al(OH)_3$		4
Alumium ICP Standar			
Aluminium oxide anhydrous	Al_2O_3		2
Aluminium oxide activated, neutral			1
Aluminum oxide　34268-8	Al_2O_3		1
Aluminum oxide powder	Al_2O_3		
Aluminium potassium sulfate 12-water	$AlK(SO_4)_2 \cdot 12H_2O$		
Aluminium standard solution	$Al(NO_3)_3$ in 1N HNO_3	A	1
Aluminium sulfate (12/1)。	$Al_2(SO_4)_3 \cdot 13\sim14H_2O$	A	1
Aluminium sulfate (12/1) 。	$Al_2(SO_4)_3 \cdot 14\sim18H_2O$		1,10
Aluminium sulfate octadecahydrate, C.P. (12/1) 。	$Al_2O_{12}S_3 \cdot 18H_2O$		
Aluminium sulfate puratronic	$Al_2(SO_4)_3$		
Amberlite IR-120(plus) lonenanstauscher Na^+			1
Amberlite XAD 7 HP			2
Ammonium acetate(11/30) 。	CH_3COONH_4		

Ammonium chloride, P.A	H_4ClN	C	P
Ammonium heptamolybdate-tetrahydrate	$(NH_4)_6Mo_7O_{24}\cdot 4H_2O$		1
Ammonium hydroxide (ammonia water)	NH_4OH	A	8
Ammonium iron(II) sufate Hexahydrate	Fe(NH4)2(SO4)2.6H2O	B,C	2
Ammonium metavanadate p.a.	H_4NO_3V		2
Ammonium molybdate(VI)	$H_{24}Mo_7N_6O_{24}\cdot 4H_2O$		
Ammonium nitrate	NH4NO3	C	1
Ammonium persulfate 98%	$H_8N_2O_8S_2$	C X2	2
Ammonium sulfate granular	$(NH_4)_2SO_4$		
Ammonium sulfate	$(NH_4)_2SO_4$		
Antimonyl potassium, Tartrate	$C_4H_4KO_7Sb\cdot 1/2H_2O$	A	2
Anthrone	$C_{14}H_{10}O$		P
L(+)-Ascorbic acid 99%	$C_6H_8O_6$		
Ascorbic acid (Vitamin C)	$C_6H_8O_6$	$C_6H_8O_6$	
B			
Bacto Yeast Extract			3
Difco Brilliant Green Bile Broth 2%			3
Bacto Tryptone			3
Barium chloride dihydrate	$BaCl_2\cdot 2H_2O$		
Barium diphenyl-aminesulfonate			2
Blue dextran			2
Boric acid p.a.	BH_3O_3	A,+1	11
Boiling stone			2
Bromocresol Green indicator			2
C			
Cadmium Standard Solution	$Cd(NO_3)_2$ in 1N-HNO_3	$Cd(NO_3)_2$ in 1N-HNO_3	
Calcium acetate monohydrate	$(CH_3COO)_2Ca\cdot H_2O$	2014.06.27	
Calcium atomic absorption standard soln AA			
Calcium Carbonate	$CaCO_3$	$CaCO_3$	3,4
Calcium Carbonate, 99% for analysis, percipitated, heavy	$CaCO_3$	A	3
Calcium Chloride, Anhydrous 95%	$CaCl_2$		
Calcium Chloride, Anhydrous	$CaCl_2$		
Calcium Chloride Dihydrate 99%	$CaCl_2\cdot 2H_2O$	A	3,P
Calcium Chloride Dihydrate 99%	$CaCl_2\cdot 2H_2O$	A	3
Calcium Chloride, fused granular	$CaCl_2$		
Calcium Fluoride (99.9%) / 99% (12/1) ◦	CaF_2	A,B	3,4,4

Calcium Hydroxide 98%	H_2CaO_2	A	3
Calcium Hydroxide •	$Ca(OH)_2$	A	3
Calcium Hydroxide •	$Ca(OH)_2$	2014.6.s27	P
Calcium Nitrate	$Ca(NO_3)_2 \cdot 4H_2O$	$Ca(NO_3)_2 \cdot 4H_2O$	
Calcium Nitrate Tetrahydrate, p.a.	$Ca(NO_3)_2 \cdot 4H_2O$		
Calcium Oxide (12/1) ◦	CaO	B	3
Calcium Sulfate Dihydrate 99%	$CaSO_4 \cdot 2H_2O$	2012/3/5	3
Calcon	$C_{20}H_{13}N_2NaO_5S$	$C_{20}H_{13}N_2NaO_5S$	3
Celite 545		X3	3,4
Cesium Nitrate 99.99%	$CsNO_3$	A,C	4
N-cetyl-N,N,N-trimethylammonium bromid ZA	$C_{19}H_{42}BrN$	B C	3
Chitosan			11
Chloroform	CHCl3		
Citric acid(11/30) ◦	$C_3H_4(OH)(COOH)_3.H_2O$		
Citric acid monohydrate	$C_6H_8O_7 \cdot H_2O$		
Citric acid monohydrate	$C_6H_8O_7 \cdot H_2O$	$C_6H_8O_7 \cdot H_2O$	
Citric acid, trisodium salt dihydrate(11/30) ◦	$C_6H_5O_7Na_3 \cdot 2H_2O$	B	4,4,P
Copper 99%	Cu	2013.01	
Cobalt(Ⅱ) chloride hexahydrate p.a.	$Cl_2Co \cdot 6H_2O$		
Copper (II) hydroxide	$Cu_{2(}OH)_2$		
Cobalt(Ⅱ) Nitrate hexahydrate	$Co(NO3)_2$		4
Cobalt(Ⅱ) Nitrate trihydrate		C	4
Copper (II) Oxide 97%	Cu_2O	2013.05	P
Copper (II) sulfate, pentahydrate	$CuSO_4 \cdot 5H_2O$	$CuSO_4 \cdot 5H_2O$	P
Copper Standard Solution	$Cu(NO_3)_2$ in 1N-HNO_3		
Copper ICP Standar			
D			
n-Decane			
Dextrose			
Diallyldimethylammonium Chloride	$(CH_2=CHCH_2)_2N(CH_3)_2$ • Cl		4
Diethylenetriamine-pentaacetic acid,98+%	C14H23N3O10	C14H23N3O1	4
N,N-Diethyl-p-phenylenediamine oxalate 96%			4
Difco™ Brilliant Green Bile Broth 2%		20120220 20120705	
Difco™ EC Medium(11/30) ◦			
Difco™ Lauryl Tryptose Broth(11/30) ◦			
Difco™ Plate Count Agar(11/30) ◦			

Dioctyl Sulfosuccinate	$C_{20}H_{37}O_{75}Na$	$C_{20}H_{37}O_{75}Na$	
Disodium Hydrogenphosphate 12, water(11/30) ◦	$Na_2HPO_4 \cdot 12H_2O$		
Dipotassium Hydrogenphosphate	K_2HPO_4		
Dodecylamine	$C_{12}H_{27}N$		6,6
Dodecylbenzene-sulfonic acid			
Dodecyl sulfate sodium salt 99% (SPS)	$C_{12}H_{25}O_4SNa$	A,A	4,4
Dodecylamine hydrochloride 99%	$C_{12}H_{27}N \bullet HCl$	A	6
Dodecylamine hydrochloride	$CH_3(CH_2)_{11}NH_2 \bullet HCl$		
E			
EDTA-2Na-Ca		A	4
Ethyl alcohol 99.5%	C_2H_5OH	B	P,P
Ethylene diamine tetraacetic acid (EDTA-2Na)	$EDTA \cdot 2Na$	A	4
Ethylene diamine tetraacetic acid	EDTA	EDTA	4
Ethylene diamine tetraacetic acid (EDTA-4Na)	$EDTA \cdot 4Na$	$EDTA \cdot 4Na$	4
Extran (sodium hydroxide solution)			
F			
Ferric Hydroxide	$Fe(OH)_3$	$Fe(OH)_3$	
Fluoride ISA Tisab 1			
Folin – Ciocalteus Phenol reagent, 2N		B 06.21	P,P
G			
Gallium (III) Nitrate hydrate 99.99989%		C	5
Gluconic acid, sodium salt, 98%			9
Glutaric dialdehyde	$C_5H_8O_2$		5
α-D+-Glucose, anhydrous, 99+%	$C_6H_{12}O_6$	$C_6H_{12}O_6$	
Gallium trichloride	5g	20120319	2
H			5
Hexamethylenetetramine	$C_6H_{12}N_4$	$C_6H_{12}N_4$	5
n-Hexane	C_6H_{14}	C_6H_{14}	8,8,8
1-Hexadecyl pyridinium chloride monohydrate	$C_{21}H_{38}ClN.H_2O$	$C_{21}H_{38}ClN.H_2O$	5
Hexadecyltrimethyl ammonium bromide 99%	$C_{19}H_{42}BrN$	A	5
Humic acid, Sodium Salt			5
Humic acid, Sodium Salt 50~60%(11/30) ◦			5
Hydrogen Peroxide(35%)	H_2O_2	H_2O_2	5,5
Hydrogen Peroxide(30%)	H2O2	H2O2	5
Hydroxylammonium chloride	$NH_2OH \cdot HCl$		
Hydroxylamine hydrochloride	NH2OH·HCl	C	5
Hydrazine sulfate p.a.	$N_2H_6SO_4$	$N_2H_6SO_4$	5
Hydrocholoric acid 37%		2012/1/31	

I			
Iron(III) chloride hexahydrate	$FeCl_3 \cdot 6H_2O$		
Iron(III) Nitrate Nonanhydrate p.a.	$FeN_3O_9 \cdot 9H_2O$	A,C	5
Iron(II) sulfate 7 hydrate	$FeSO_4 \cdot 7H_2O$	B	5
Iron(II) sulfate 7 hydrate	$FeSO_4 \cdot 7H_2O$	$FeSO_4 \cdot 7H$	5
Indium trichloride		20120326	5
K			
Kaolin			16
Kaolin			16,P
L			
Lanthanum oxide (12/1) ◦	La_2O_3		
Lauric acid 95%	$C_{12}H_{24}O_2$		
Lead Standard Solution			
Lithium hydroxide monohydrate 56%	LiOH(HLiO*H_2O)		5
M			
Magnafloc 156			22
Magnafloc LT 31			
Magnesium chloride hexahydrate 99%	$Cl_2Mg \cdot 6H_2O$	A,C	6,7,7
Magnesium chloride anhydrous	$MgCl_2$		6,6,6
Manganese chloride tetrahydrate	MnCl2.4(H2O)	C	6
Magnesium Hydroxide	$Mg(OH)_2$	$Mg(OH)_2$	7
Magnesium nitrate hexahydrate, 99%, for analysis	$MgN_2O_6.6H_2O$	C	7
Magnesium oxide, powder 98%	MgO		6,P
Magnesium Phosphate Octahydrate	$Mg_3(PO_4)_2.8H_2O$	$Mg_3(PO_4)_2.8H_2O$	6
Magnesium sulfate heptahydrate p.a. •	$MgSO_4 \cdot 7H_2O$		
Magnesium sulfate heptahydrate •	$MgSO_4 \cdot 7H_2O$		6
Manganese dioxide, Granular	MnO_2	MnO_2	6
Manganese (IV) oxide, powder	MnO_2	MnO_2	6
Manganese sulfate monohydrate	$MnSO_4 \cdot H_2O$	$MnSO_4 \cdot H_2O$	6
Mercury (II) Chloride p.a.	Cl_2Hg	Cl_2Hg	6
Mercury (II) iodide ,red p.a.	HgI_2		
Methanesulfonic acid	CH_4O_3S		7
Methanol			6
Methyl alcohol, 99.9%	CH_3OH	CH_3OH	8
Methyl alcohol, 99.9%	CH_3OH		
Methyl Red, indicator p.a.			6
Methyl Orange, indicator	$C_{14}H_{14}N_3NaO_3S$	$C_{14}H_{14}N_3NaO_3S$	6
Methylglycol chitosan, Minimum 70%			6,6,6,6

Montmorillonite K10		A	7,7
Montmorillonite KSF			7,7
N			
Natriumfluorid	NaF		
Nickel Standard Solution	$NiCl_2$ in 1 N HCl		
Nitric acid			9
O			
Orthophosphorique acid			9
Oxalic acid dihydrate	$C_2H_2O_4.2H_2O$		7
P			
Palladium(II) Chloride 59%	$PdCl_2$	2014.08.04	7
1,10-Phenanthroline 99%			6
1,10-Phenanthroline monohydrate	$C_{12}H_8N_2 \cdot H_2O$		
1,10-Phenanthroline monohydrate	$C_{12}H_8N_2 \cdot H_2O$		7
1,10-Phenanthroline monohydrate	$C_{12}H_8N_2 \cdot H_2O$		6
Phenol 99%	C_6H_6O		8,P
Phenolphthalein	$C_{20}H_{14}O_4$	$C_{20}H_{14}O_4$	8
Phosphoric acid 85 wt%	H_3O_4P	H_3O_4P	9
Phosphorus Standard Solution	KH_2PO_4 in H_2O		
Phosphor ICP Standar			
Poly Aluminium Chloride			23
Poly(acrylamide) 50 wt% solution in water	$[-CH_2CH(CONH_2)-]n$	$[-CH_2CH(CONH_2)-]n$	23
Poly(acrylamide)非離子 (non-ionic)			23
Poly(acrylamide-co-acrylic acid)	$[-CH_2CH(CONH_2)-]x[-CH_2-CH(CO_2H)-]y$	$[-CH_2CH(CONH_2)-]x[-CH_2-CH(CO$	22
Poly(acrylic acid)	$[-CH_2CH(CO_2H)-]n$	A	22,23,23
Poly(acrylic acid), 25 wt% solution in water			22
Poly(acrylic acid), 63 wt% solution in water			23
Poly (diallyldimethylammonium chloride)20 wt% high molecular weight	M:10~20,20~35,40~50 萬	M:10~20,20~35,40~50 萬	23,23
Poly (diallyldimethylammonium chloride)20 wt% medium molecular weight	M:10~20,20~35,40~50 萬	M:10~20,20~35,40~50 萬	23
Poly (diallyldimethylammonium chloride)20 wt% low molecular weight	M:10~20,20~35,40~50 萬	M:10~20,20~35,40~50 萬	23
Poly(ethylene glycol)	MW 1.5K 6K 10K 20K 0.4K		22
Poly(ethylenimine) 50wt% aq solution			22
Polyoxyethylenlaurylether(Brij 35)			22,22

Polyvinyl sulfur acid potassium salt 90%			6
Potassium bromide	KBr		
Potassium bromide, 99+%	KBr	KBr	
Potassium chloride p.a.	KCl	C	8
Potassium dihydrogen phosphate	H_2KPO_4	H_2KPO_4	6
Potassium dichromate GR ·	$K_2Cr_2O_7$		
Potassium dichromate ·	$K_2Cr_2O_7$	$K_2Cr_2O_7$	7
Potassium hexachloroplatinate (Ⅳ)ca.40%Pt	Cl_6K_2Pt	C	6,6,6,6
Potassium hydrogen phthalate(12/1) ◦	$C_8H_5KO_4$		
Potassium hydrogen phthalate, 99%, acidimetric standard	$C_8H_5KO_4$	$C_8H_5KO_4$	P
Potassium hydroxide	KOH	KOH	8,P
Potassium iodide, 99%	IK	IK	
Potassium Nitrate	KNO_3	KNO_3	P
Potassium persulfate (12/1) ◦	$K_2S_2O_8$		
Potassium Permanganate, 98% extra pure	$KMnO_4$	$KMnO_4$	8
Potassium phosphate, monobasic, p.a.	KH_2PO_4	KH_2PO_4	8
Potassium phosphate, dibasic, 99%, for analysis, anhydrous	HK_2O_4P	HK_2O_4P	8
Potassium sodium tartrate tetrahydrate	$C_4H_4KNaO_6 \cdot 4H_2O$	HK_2O_4P	P
Potassium sulfate anhydrous powder	K_2SO_4	K_2SO_4	7
2,6-Pyridine-dicarboxylic acid 99%	$C_7H_5NO_4$	$C_7H_5NO_4$	6,6
poly (vinyl alcohol) 88% M.W.20000-30000		2012/6/26	23
poly (vinyl alcohol) 95% M.W.95000			23
poly (vinyl alcohol) M.W. 2000		2012/4/16	22
R			
Redox Standard 220mV			9
S			
Saponin		B	10
Sephadex G-75			10,10
Silberoxid	Ag_2O		
Silica, colloidal			10
Silica gel			10
Silica gel for column chromatography			P
Silica gel, Medium Granular (Blue)			10
Sodium acetate anhydrous	CH_3COONa		

Sodium arsenate dibasic	$Na_2HAsO_4 \cdot 7H_2O$		
Sodium arsenaite			10
Sodium azide 99%	NaN_3	NaN_3	10,10
Sodium bicarbonate p.a. (12/1) ◦	$CHNaO_3$		10,P
Sodium bisulfite, powder, p.a.	$HNaO_3S$	$HNaO_3S$	11
Sodium borate, 10-hydrate crystal	$Na_2B_4O_7 \cdot 10H_2O$	2012/1/31	P
Sodium borohydride 98%	H_4BNa		P, P
Sodium carbonate •	Na_2CO_3	B	P
Sodium carbonate anhydrous •	Na_2CO_3	Na_2CO_3	11
Sodium chloride p.a.	NaCl		11,P
Sodium dihydrogenphosphate dihydrate	$NaH_2PO_4 \cdot 2H_2O$		
Sodium disulfite, p.a.	$Na_2S_2O_5$	$Na_2S_2O_5$	11
Sodium fluoride	NaF	NaF	10,10,10
Sodium fluoride, 97%	NaF	NaF	10
Sodium fluoalummium	Na_2AlF_6	NaF	11
Sodium gluconate	$C_6H_{11}NaO_7$		
Sodium hexametaphosphate (12/1) ◦	$(NaPO_3)_6$		
Sodium hydrogen carbonate	$NaHCO_3$	2012/3/5 A	9
Sodium hydroxide	NaOH	NaOH	11,P,P
Sodium hypochlorite	NaOCl	B	11,P
Sodium metasilicate, nonahydrate 44-47.5%	$Na_2SiO_3 \cdot 9H_2O$	$Na_2SiO_3 \cdot 9H_2O$	11
Sodium molybate dihydrate	$Na_2MoO_4 \cdot 2H_2O$	2012/2/3	10
Sodium nitrate •	$NaNO_3$	$NaNO_3$	D
Sodium Nitrate • 99%	$NaNO_3$	B 2014.04.24	10,10,P,P
Sodium Nitrite, 98.5%, for analysis	$NNaO_2$	C	10
Strontium Nitrate		2013.09.30 C	10
Sodium nitroprusside	$Na_2Fe(CN)_5(NO) \cdot 2H_2O$	A	P
Sodium nitroferricyanide dehydrate		2013.01	P
Sodium Oleate	$_4(CH_2)_7CH{:}VH(CH_2)_7 \cdot CO$		11
Sodium Peroxodisulfate	$Na_2S_2O_8$	B	11
Sodium Perchlorate, ACS	CAS 7601-89-0		10
Sodium persulfate, 98%	$Na_2O_8S_2$	2014.6.30 04.21	11,P
Sodium Phosphate dibasic heptahydrate, p.a.	$Na_2HPO_4 \cdot 7H_2O$		
Sodium Pyrophosphate 10 Hydrate (12/1) ◦	$Na_4P_2O_7 \cdot 10H_2O$		
Sodium Sulfite, anhydrous, p.a.	Na_2SO_3	2014.6.30 05.02	11,11,P,P

Sodium Sulfite, anhydrous, p.a.	Na_2SO_4		9
Sodium thiosulfate penthydrate	$Na_2S_2O_3 \cdot 5H_2O$		
ICP Standard			
Standard solution, Al			12
Standard solution, Ca	Ca		12
Standard solution, Cd	Cd	Cd	12
Standard solution, Cr	Cr	Cr	12
Standard solution, Cu	Cu	Cu	12
Standard solution, Fe	Fe	Fe	12
Standard solution, Ga			12
Standard solution, In	In		12
Standard solution, K			12
Standard solution, La			12
Standard solution, Mg			12
Standard solution, Mn	Mn	Mn	12
Standard solution, Mo			12
Standard solution, Ni	Ni	Ni	12
Standard solution, Na	Na	Ni	12
Standard solution, P			12
Standard solution, Pb	Pb	Pb	12
Standard solution, Pd		2014.7.22	12
Standard solution, Sr			12
Standard solution, Si			12
Standard solution, Te			12
Standard solution, Ta			12
Standard solution, Zn	Zn	Zn	12
Multi – As, Be, Cd, Cr(VI), Hg, Ni, Pb, Se, Tl			
Multi – Ag, Al, B, Ba, Bi, Ca, Cd, Co, Cr, Cu, Fe, Ga, In, K, Li, Mg, Ma, Na, Ni, Pb, Sr, Tl, Zn			
IC Standard			
Standard solution, NH^{4+}			12
Standard solution, F^-		2013.09.27	12
Standard solution, PO_4^{2-}			12
Standard solution, SO_4^{2-}			12
Superfloc 581 C			
Sulfuric Acid (>95%)		B	3
Sulfanilic acid azochromotrop 80%			18
T			

Talcum			
Tetrahydrofuran	C4H8O		
Titriplex I (12/1) ◦	$C_6H_9NO_6$		16
Toluidinblau O(Toluidine Blue N)			6
Toluidine Blue O	$C_{15}H_{16}ClN_3S$	$C_{15}H_{16}ClN_3S$	6
Titanium oxide 325 mesh	TiO_2		11
Tran-1,2-diamino-cyclohexane-N,N,N',N'-Tetraacetic acid monohydrate 98% (CDTA)			11,11
Tricalcium Bis (orthophosphate)	$Ca_3(PO_4)_2$		
Trichloroacetic acid	CCl_3COOH	CCl_3COOH	6
Trietanilamine	C6H15NO3	C6H15NO3	16
Triton X-100	$C_{34}H_{62}O_{11}$	$C_{34}H_{62}O_{11}$	16
Tris(1,10-phenanthroline) iron(Ⅱ) sulfate			11
Z			
Zinc Standard solution	$Zn(NO_3)_2$ in 1 N HNO_3		
Zinc Sulfate, Heptahydrate(11/30) ◦	$ZnSO_4.7H_2O$		
Zirconyl chloride octahydrate 98%			16
Technical Buffer pH = 4.01			
Technical Buffer pH = 7.00			
Technical Buffer pH = 10.00			